

## 1 Introduction

- Agricultural fire emissions can have a large impact on atmospheric composition and air quality on regional scales.
- Chemical and physical transformations of primary emissions can lead to significant changes in gaseous and particulate phase compositions of the smoke.
- Emissions and smoke chemistry are not well characterized.

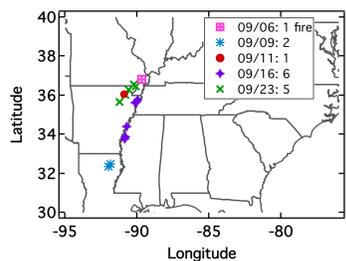
## 2 Objectives

- To quantify emissions of trace gases and fine particles from 15 agricultural fires.
- To study the evolution of NO<sub>y</sub> species (PAN, NO<sub>x</sub>, HNO<sub>3</sub>, nitrate), O<sub>3</sub>, organic aerosol (OA), and brown carbon (BrC) in fire plumes.

## 3 Aircraft Instrumentation

Gas	SO <sub>2</sub> , HCl, PAN HCN, Hydroxyacetone, C <sub>5</sub> O <sub>3</sub> H <sub>8</sub> , HNO <sub>3</sub> VOC and OVOC NO <sub>x</sub> , NO <sub>y</sub> , O <sub>3</sub> CO <sub>2</sub> CO CH <sub>2</sub> O	GaTech CIMS Caltech CIMS Innsbruck PTR-MS NOAA Chemiluminescence NASA AVOCET NASA DACOM Laser-induced fluorescence
Aerosol	BC SO <sub>4</sub> , NO <sub>3</sub> , NH <sub>4</sub> , Cl, OA	NOAA SP2 CU HR-ToF-AMS
Optical	Particle absorption coefficients	NASA PSAP

## 4 Fires Sampled



Vegetation: rice straw

## 5 Analysis Methods

- Normalized excess mixing ratio (NEMR) was used to calculate emission factors and to study evolution

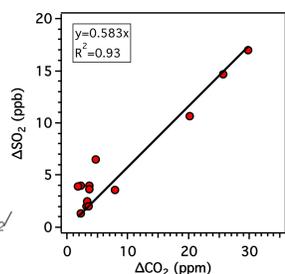
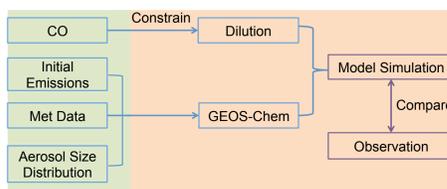


Figure 1. Emission ratio plot of ΔSO<sub>2</sub>/ΔCO<sub>2</sub> from fire #1 on Sept 11.

- A Lagrangian box model was used for modeling smoke chemistry



## 5 Emission Factors

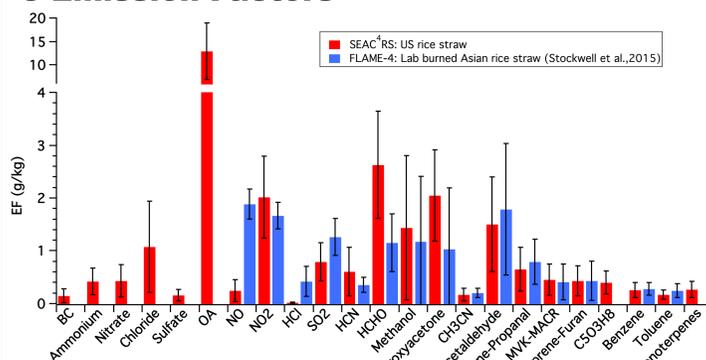


Figure 2. Emission factors (g kg<sup>-1</sup>) measured during SEAC<sup>4</sup>RS and FLAME-4.

- In general, the average EFs of SEAC<sup>4</sup>RS and those predicted from the FLAME-4 EF vs. MCE plot for the overlapping species are shown to agree well

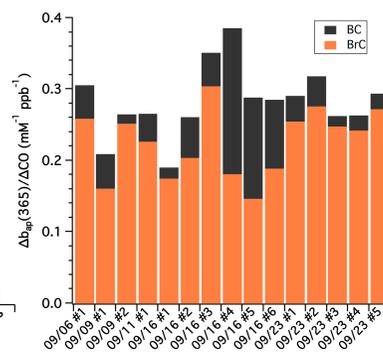
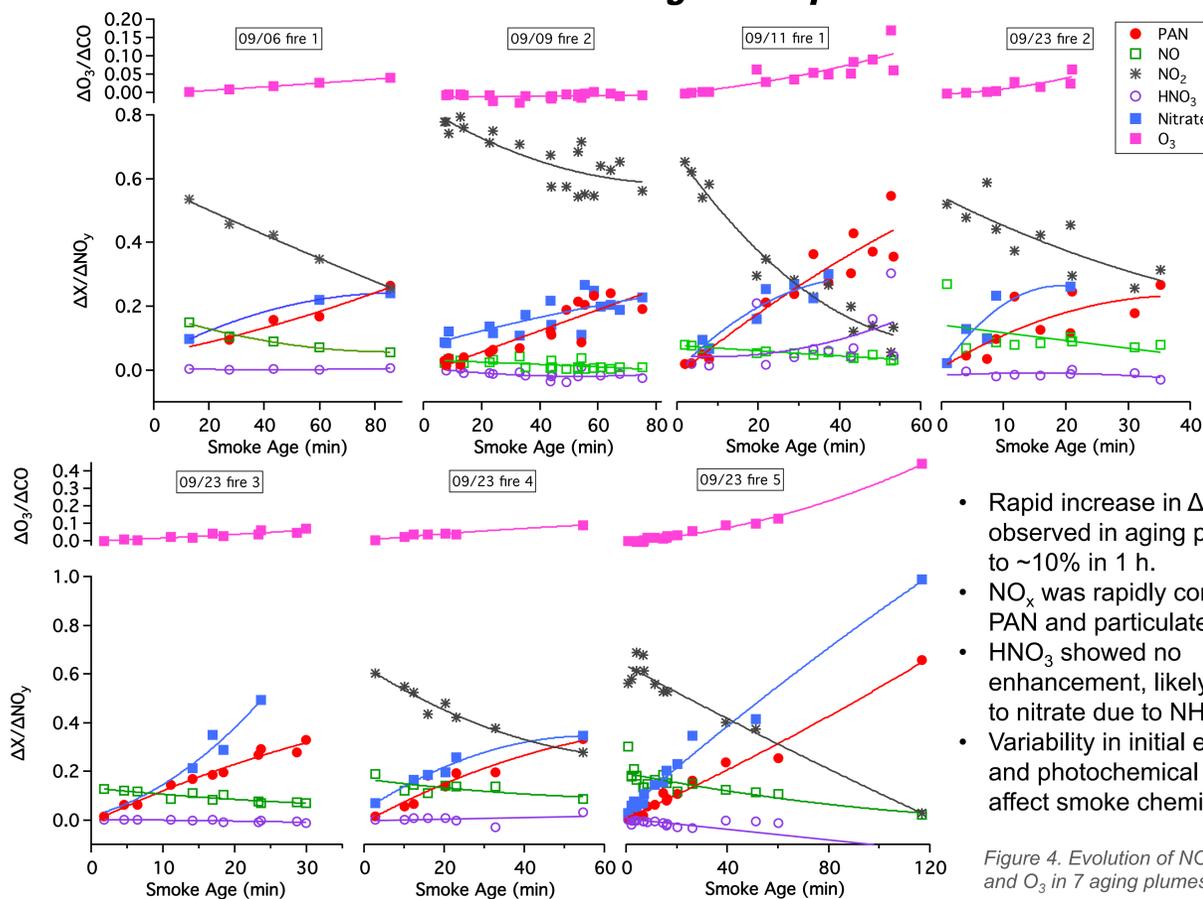


Figure 3. Total light absorption coefficient at 365 nm with BrC and BC contributions.

- PSAP data indicate that BrC is ubiquitous in agricultural fire plumes

- Absorption Ångström Exponent (AAE) inferred from PSAP absorption at 470 and 532 nm
- BrC absorption at 365 nm derived from aerosol AAE and a BC AAE of 1

## 6 Evolution of Ozone and Reactive Nitrogen Compounds



- Rapid increase in ΔO<sub>3</sub>/ΔCO observed in aging plumes, up to ~10% in 1 h.
- NO<sub>x</sub> was rapidly converted to PAN and particulate nitrate
- HNO<sub>3</sub> showed no enhancement, likely converted to nitrate due to NH<sub>3</sub> in plumes
- Variability in initial emissions and photochemical processing affect smoke chemistry

Figure 4. Evolution of NO<sub>y</sub> species and O<sub>3</sub> in 7 aging plumes.

## 7 Evolution of OA and BrC

- No significant difference downwind in the first hour of aging
- Increase in O:C and decrease in H:C observed in some fires

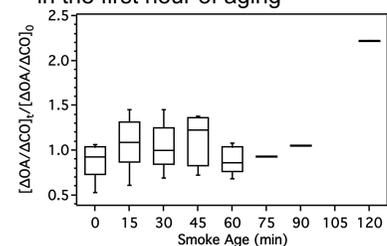


Figure 5. Whisker plot of OA evolution.

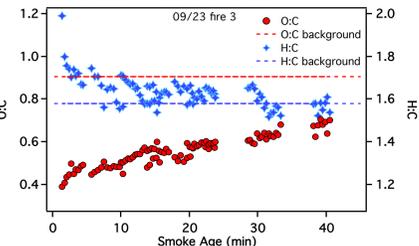


Figure 6. Change of O:C and H:C ratios.

- Very slight increase in BrC absorption

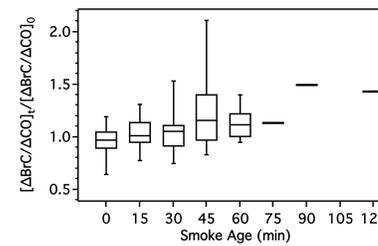
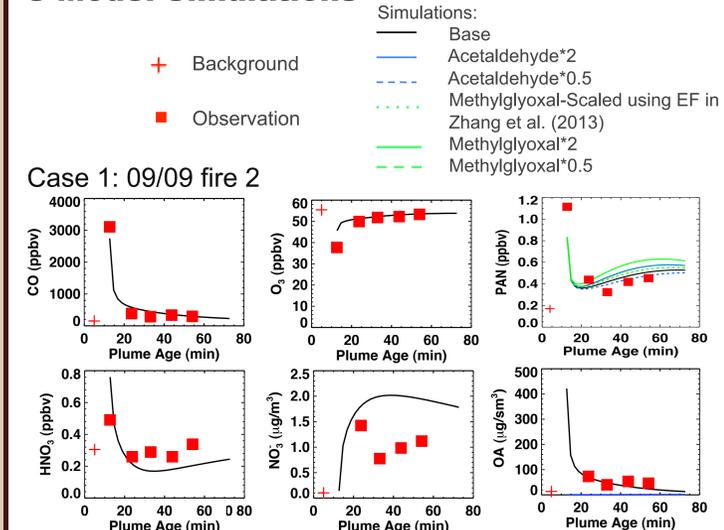
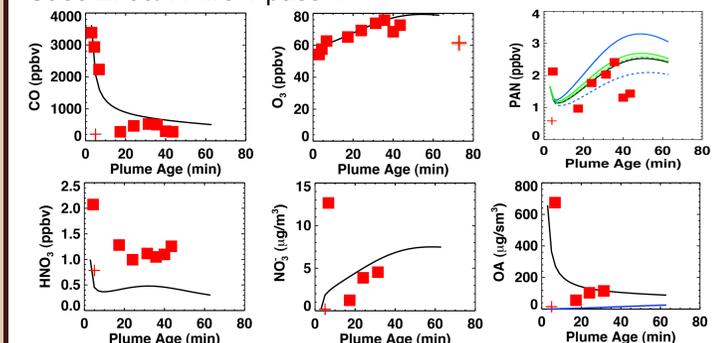


Figure 7. Whisker plot of BrC abs evolution.

## 8 Model Simulations



### Case 2: 09/11 fire 1 pass 1



- Modeled O<sub>3</sub> and PAN agree with the observations within ~30%.
- PAN is sensitive to OVOCs, e.g., acetaldehyde and methylglyoxal.
- The HNO<sub>3</sub>-NO<sub>3</sub> system is more complicated than the model can represent. For example, while the model assumes no initial emission of nitrate, AMS data show the opposite for Case 2.
- Modeled SOA production is small compared to the OA variability.

## 9 Conclusions

- Emission factors for a set of gaseous and particulate species were measured.
- BrC was produced in agricultural fires.
- As the plumes aged, fast enhancement of O<sub>3</sub> and conversion of NO<sub>x</sub> to PAN and nitrate were observed.
- The ratio of OA to CO and that of b<sub>ap, BrC</sub> to CO were not significantly different downwind during the first hour of aging. However, the increase of O:C ratio indicates ongoing photo-oxidation.
- The model reasonably simulates dilution and chemical processes in aged fire plumes, especially for O<sub>3</sub> and PAN chemistry.

## Acknowledgement

This work was funded by NASA. We thank the SEAC<sup>4</sup>RS science team and the DC-8 crew.